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RESEARCH ARTICLE

AN APPRAISAL OF UNTREATED AND TREATED WATER IN EASTERN DELHI REGION USING GEOGRAPHIC INFORMATION SCIENCE

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ABSTRACT

From last three decades, industrial revolution in India on boom. And due to this lot of water quality problem arises. Industries are directly drain the effluents in to the water bodies which degrading the water quality and also life. In Delhi region, Yamuna river is the major sink for the industrial drains. In this context, this paper displays a comparative study of untreated water and treated water of Yamuna River for the water quality assessment.

Key words:

Water quality, Untreated water,
Treated water, Yamuna River.

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INTRODUCTION

In India, riverine resources contain 113 river basins out of which 14 are majors and remains are minor. (Sharma *et al.*, 2009; Sengupta, 2006; Karim and Veizer, 2000; Galy *et al.*, 1999; Rengarajan *et al.*, 2009; Sinha *et al.*, 2013). These 14 major river basin accounts for 85% of the total surface flow and 83% of the total area of the basin and contribute and cover 80% of the total length. (Jacob *et al.*, 2001; Abbaspour *et al.*, 2007; Shiklomanov *et al.*, 2003; Amarasinghe *et al.*, 2005; Gupta *et al.*, 2005; Sharma *et al.*, 2009). Due to industrialisation, environmental pollution became a conundrum for human race (Pal, 2011; Taylor, 1996; Friedmann, 2005; Carey, 2007). Lots of research has been carried out by different authors in assessing the water quality profile of rivers using different pollution stress parameters. (Sharma *et al.*, 2009; Hamouda *et al.*, 2009; Baudrimont *et al.*, 1999; Servais *et al.*, 2007; Mouillot *et al.*, 2006). Yamuna river pollution has been described taking simple parameters viz. Sulphates, fluoride, nitrate, TDS, alkalinity, chloride, total hardness (Singh *et al.*, 2012; Tyagi *et al.*, 2008; SINGH, 2006; Goel, 2006). The major tool of pollute on profile studies is the water quality index (WQI) (Bordalo *et al.*, 2006; Taner *et al.*, 2011; Sharma *et al.*, 2009; Reza, and Singh, 2010; Lumb *et al.*, 2011). An overall index of pollution for surface water based on average value of all the pollution sub indices gives a general classification scheme in the Indian context. (Sargaonkar and Deshpande, 2003; Boyacioglu, 2007; Sharma *et al.*, 2009; Alobaidy *et al.*, 2010; Sarkar *et al.*, 2006; Alobaidy *et al.*, 2010; Dede *et al.*, 2013). Contamination of drinking water due to exposure to biological and chemical pollutants is a major cause of illness and mortality. (Mokdad *et al.*, 2004; Azevedo *et al.*, 2002; Pimentel *et al.*, 1998; Järup, 2003; Nash, 1993; Metzger *et al.*, 1995). The change in pH values of water bodies have been reported to affect aquatic life and alter toxicity of other pollutant in one form or the other (Pal and Vimala, 2012; Federici *et al.*, 2007; Novotny, 1995; Ogunfowokan *et al.*, 2005; Knutzen, 1981; Cooper, 1993).

These chemical compounds and salts present in the river Yamuna are hazardous for the health of humans. (Agrawal *et al.*, 2010; Kumar *et al.*, 2010; Ohe *et al.*, 2004; Gopal, and Agarwal, 2003; Lalwani *et al.*, 2005). There are different kinds of diseases triggered by these kinds of chemicals and salts. (Sasahara *et al.*, 2007; Morris *et al.*, 2008; Baillie-Hamilton, 2005; Joslin, 1849) Sulphates and fluorides are causing various skin irritation and various type of ortho and Neuro diseases. (Simate *et al.*, 2012; Murlidhar *et al.*, 2009; Walmsley and Moore, 1999). Nitrates causing blue baby syndromes and skin irritations. (Eneh, 2011; Dorado and Agbisit, 2001; Fisher and Fisher, 2001; Eneh, 2011; Sawian *et al.*, 2008). The other factors like TDS, alkalinity, chlorides and total hardness makes the water unfit for drinking and consuming (Asadi *et al.*, 2008; Murhekar Gopalkrushna, 2011; Goel, 2006; Singh *et al.*, 2003; Manjunatha *et al.*, 2012) In developing countries, distribution network has become a point at which contamination frequently occurs to unacceptably high levels, posing a threat to public health. (Vairavamoorthy *et al.*, 2007; Vairavamoorthy *et al.*, 2007; Srivastava, 2008; Lerner, 2004). In the present research work, the analysis and estimation of sulphates, fluorides, nitrates, TDS, alkalinity, chloride and total hardness have been done of the Yamuna river water near Wazirabad, New Delhi. The work is performed to discuss the suitability of groundwater for human consumption based on the water quality index.

Study Area

Wazirabad is located in the Delhi region (28° 42' 15.5" N; 77° 14' 03.9" E) near Yamuna River. Various types of industrial effluents and agricultural runoffs of Delhi drained in to the Yamuna River in Wazirabad region. The Yamuna River flowing in a southerly direction in the eastern part of the Union Territory of Delhi is the only perennial river in the area. In Wazirabad region, the water treatment plant situated where water has been treated and further distributed in Delhi. But available water treatment facilities are not adequate to remove all the pollutants. Consequently, the Yamuna leaves Delhi as a sewer, laden with the city's biological and chemical wastes. Downstream, at Agra, this becomes the main municipal drinking

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water source. Here, too, the existing treatment facilities are inadequate to treat the pollutants. Thus, consumers in Delhi and Agra ingest unknown amounts of toxic pesticide residues each time they drink water (CSE, 2008). Collection of untreated water and treated water of Wazirabad region has done to study the seven parameters that mostly affect the human health. The sample points were indicated through the following map (Figure 1) and overlaid on Land sat TM satellite image which was acquired on 08-05-2011. The image encompassed by blue, green and red bands. In map sampling point 1 are untreated water (SP 1 W RW) and sampling point 2 are treated water (SP 2 W TW) in Wazirabad areas.

1995) and Bureau of Indian Standards (BIS, 1991) which are represented in table. (Table 1)

Table 1. Analytical method used for water quality analysis

Parameter	Analytical method
Total alkalinity	Acid titration method
Total hardness	EDTA titration method
TDS	Total Solids Dried at 103–105°C
Chloride	Argentometric titration method
Sulphate	Spectrophotometric method
Fluoride	SPADNS spectrophotometric method
Nitrate	Spectrophotometric method

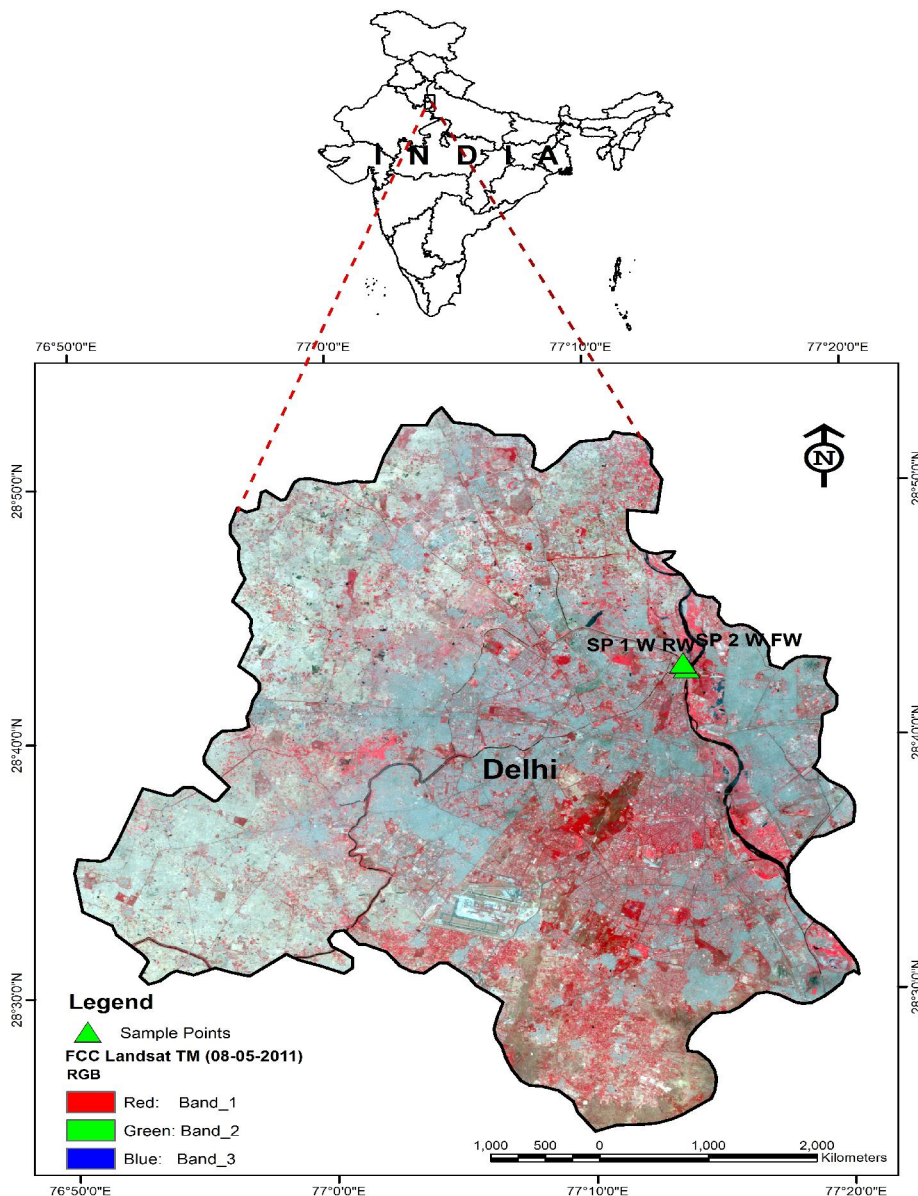


Fig.1. Location map of study site

Note: (SP 1 W RW) Sampling Point-1 Wazirabad untreated water
(SP 2 W TW) Sampling Point-2 Wazirabad treated water

MATERIALS AND METHOD

Untreated water and treated water samples were collected from selected areas up to 9 Months (January 2012 to September 2012) sampling done twice in a month that is supply for drinking purposes. The water samples were collected in pre-cleaned plastic containers. The chemical constituents were analyzed in laboratory using standard methods determined by American Public Health Association (APHA,

In drinking water quality assessment, priority should be given to those substances which are known to be important to health, portability and which are known to be present in significant concentrations in the water source (World Health Organization (WHO), 2006a). The various water quality analyses were carried out following the 20th edition of the Standard Methods for Examination of Water and Wastewater, published by APHA, AWWA, WEF (1998). The Bureau of Indian Standards (BIS) has published the Drinking Water Quality Specifications (IS-10500:1991) these water quality guidelines

Table 2. Effect of Water Quality Parameters for Drinking Purpose

Parameters	Prescribed limits IS:10500, 1991		Probable effects
	Desirable limits	Permissible limit	
Hardness as CaCO ₃ (mg/L)	300	600	Affects water supply system (scaling), Excessive soap consumption, and calcification of arteries. There is no conclusive proof but it may cause urinary concretions, diseases of kidney or bladder and stomach disorder
Chloride (mg/L)	250	1000	Injurious to some people suffering from diseases of heart or kidneys. Taste, indigestion, corrosion and palatability are affected.
Total Dissolved Solids (mg/L) TDS	500	2000	Palatability decreases and may cause gastro-intestinal irritation in human may have laxative effect particularly upon transits and corrosion may damage water system.
Sulphate (SO ₄) mg/L	200	400	Causes gastro intestinal irritation along with Mg or Na can have a cathartic effect on users, concentration more than 750 mg/L may have laxative effect along with Magnesium.
Nitrate (NO ₃) (mg/L)	45	100	Cause infact methaemoglobinaemia (blue babies) at very high concentration, causes gastric cancer and affects adversely central nervous system and cardiovascular system.
Fluoride (F) (mg/L)	1	1.5	Reduces dental carries, very high concentration may cause crippling skeletal fluorosis.
Alkalinity mg/L CaCO ₃	200	600	Impart distinctly unpleasant taste may be deleterious to human being in presence of high pH, hardness and total dissolved solids.

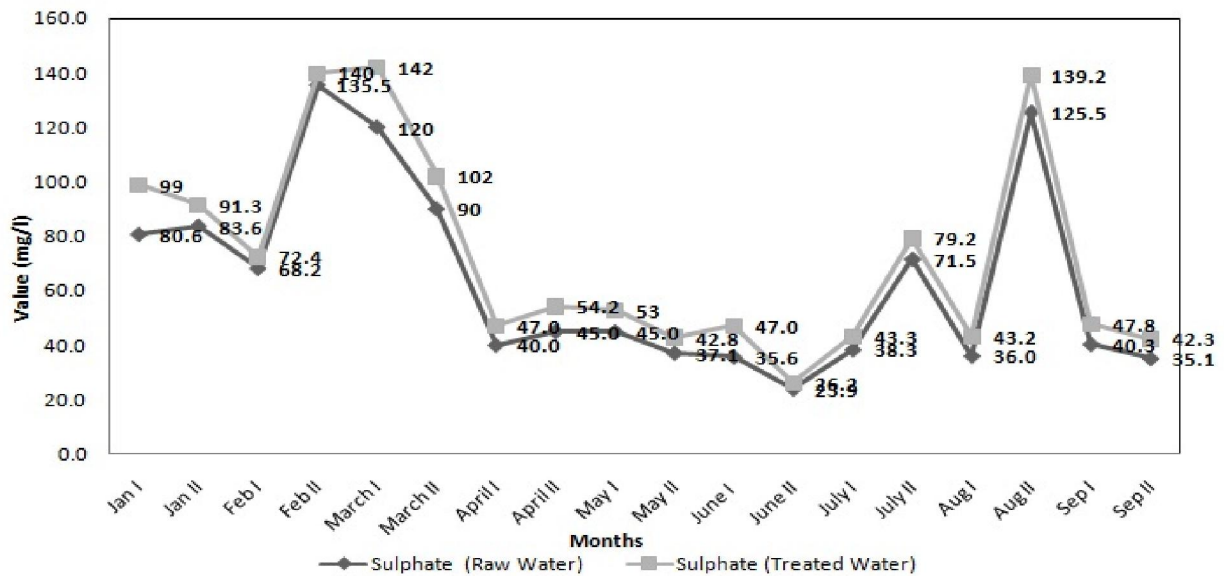


Fig.2. Variation of Sulphate in different months

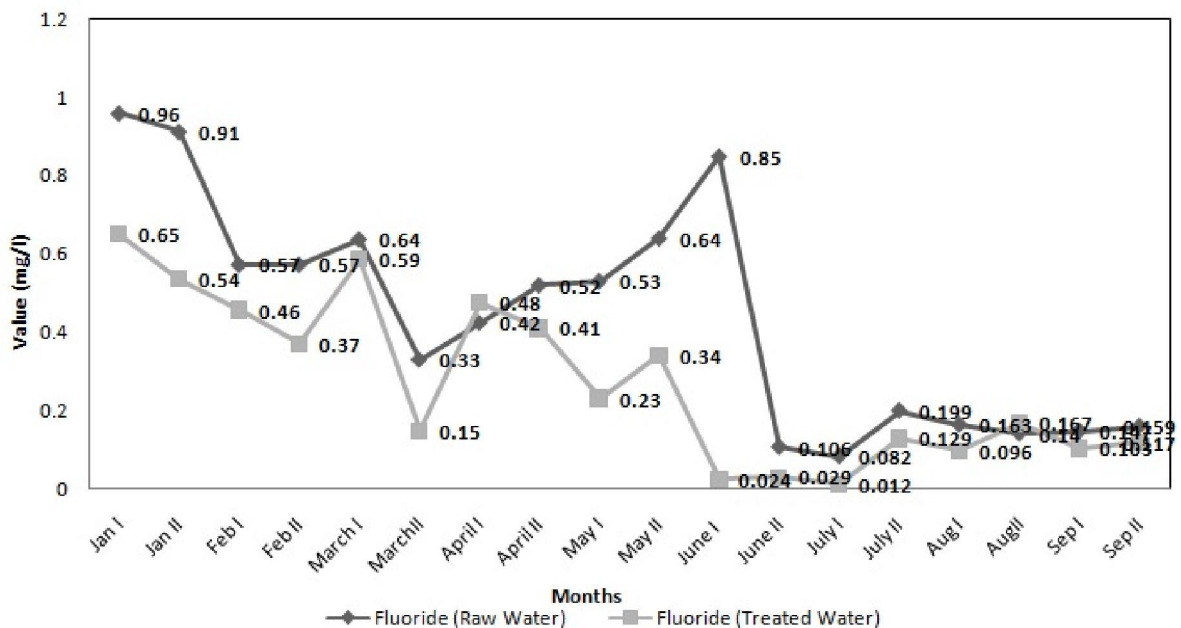


Fig.3. Variation of Fluoride in different months

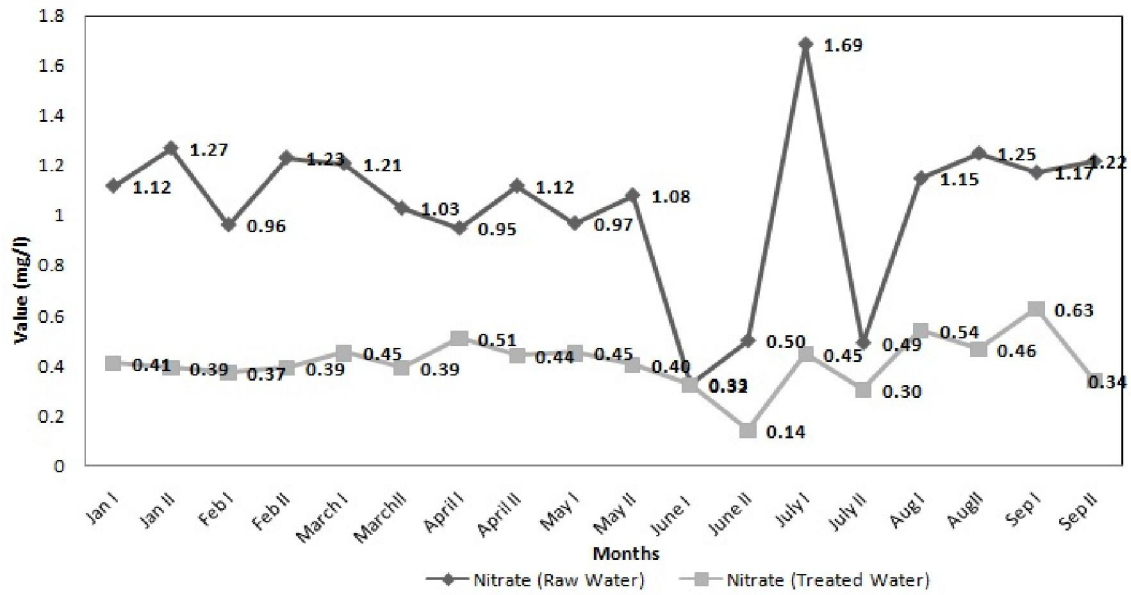


Fig.4. Variation of Nitrate in different months

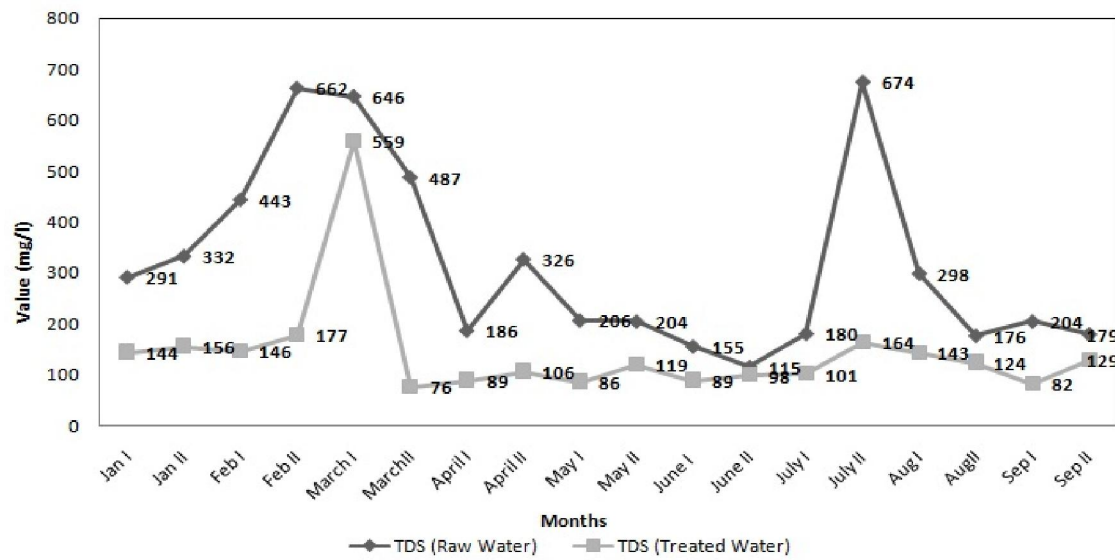


Fig.5. Variation of TDS in different months

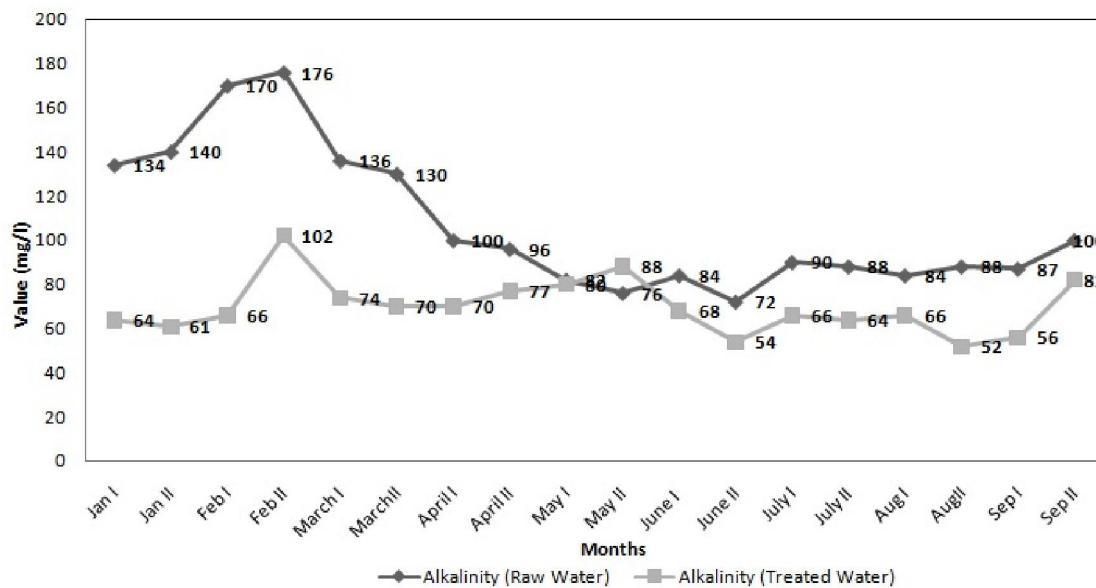


Fig.6. Variation of Alkalinity in different months

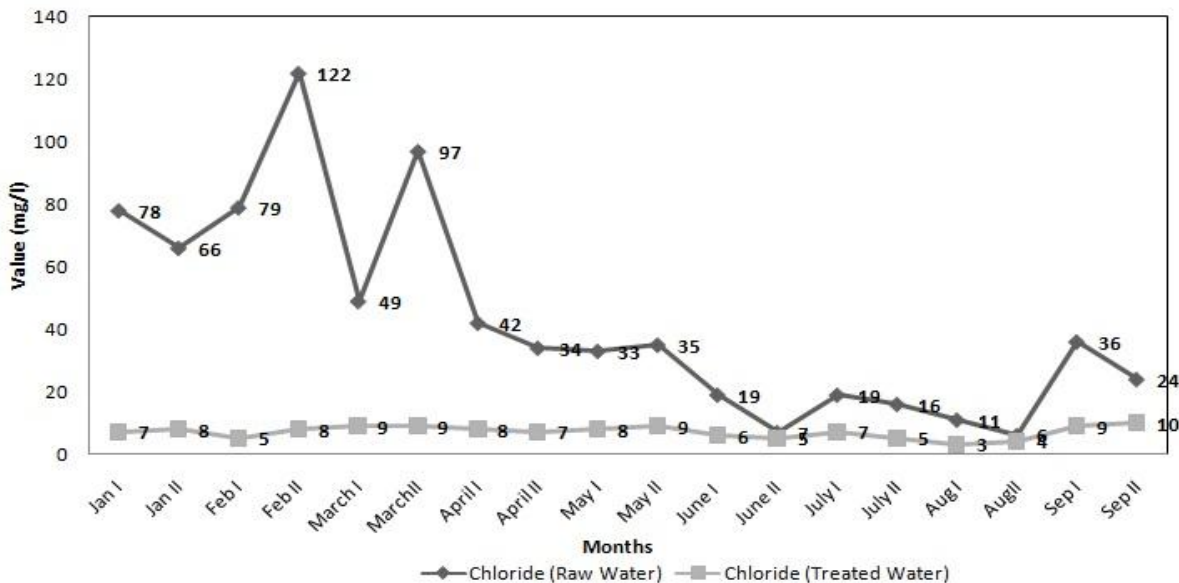


Fig.7. Variation of Chloride in different months

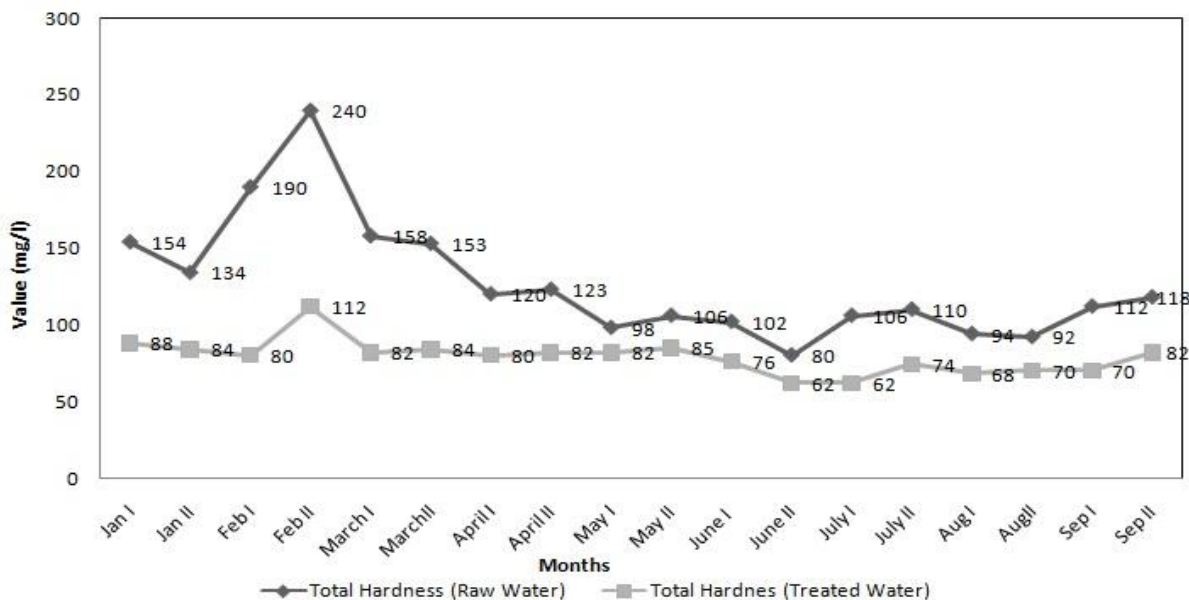


Fig.8. Variation of Total hardness in different months

and specifications have been employed for analysis of results and preliminary assessment of the water quality of river water. The water quality is adjudged based on physical (color, odor, pH, electrical conductivity, turbidity, total dissolved solids, etc.), chemical (chloride, sulphate, carbonate, bicarbonate, nitrate, nitrite, hardness, alkalinity, sodium, potassium, calcium, magnesium, boron, fluoride, iron, manganese, arsenic, copper, mercury, nickel, cadmium, aluminum, chromium, zinc, lead, etc.), bacteriological (total coliforms, fecal coliforms, fecal streptococcus, etc.), biological (DO, BOD, COD, TOC, etc.) and radiological characteristics.

RESULTS AND DISCUSSION

In the study we found that all the parameters were variable in both untreated water and treated water. Figure-2 shows the variation of sulphate in Treated water and untreated water, in treated water sulphate is high in month of February II (140 mg/L) and March I (142mg/L) as compare to untreated water. Sulphate is higher because of mixture of alum for treated of untreated water. In fluoride (Figure-3), it is high in month of January I (0.96 mg/L) and January II (0.91 mg/L) in the untreated water as compare to treated water.

Indian Standard 10500-1991 gives the Desirable limit (1.0 mg/L) and Permissible limit (1.5 mg/L). The important source of nitrate is biological oxidation of organic nitrogenous substances which come in sewage and industrial wastes, domestic sewage contains very high amount of nitrogenous compounds. Figure-4 show the nitrate concentration is high in July I (1.69 mg/L) in untreated water and in treated water concentration of nitrate is lower in June II (0.14 mg/L). Figure-5 show the different between TDS in untreated water and treated water, in untreated water the TDS are high in month of July II (674 mg/L) and also show the higher concentration in February II (662 mg/L) and in treated water the TDS are higher in March I (559 mg/L). TDS denote mainly the various kinds of minerals present in water that is untreated water or treated water. Alkalinity of surface water is primarily a function of carbonate, bicarbonate and hydroxide content. Figure-6 show Alkalinity of untreated water is higher in month of February II (176 mg/L) and in treated water it is also higher in February II (102 mg/L). Chloride is major inorganic anions in water and waste water, the salty taste produced in potable water is due to presence of chloride ions. Figure-7 show the concentration of chloride of untreated water and treated water and chloride are high in month of February II (122 mg/L). Total hardness generally divided by

calcium and magnesium ions i.e. come in water. Figure-8 shows the different concentration of total hardness in different month of untreated water and treated water, in the untreated water the total hardness is higher in the month February II (240 mg/L).

Conclusion

Present study comprises the seven different parameters and it is depicted that the treated and untreated water of Wazirabad region have variations according to the months. In different months, the value of these parameters have changed and also found out higher than the permissible limits (Sulphate 142 mg/l (March); Flouride 0.65 mg/l (January); Nitrate 0.63 mg/l (September); Total Dissolved Solids 559 mg/l (March); Alkalinity 102 mg/l (February); Chloride 10 mg/l (September); Total Hardness 112 mg/l (February). The analysis reveals that the water of the area needs more degree of treatment before consumption and it also needs to be protected from the perils of contamination.

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